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## HYDROGEN MARKETS & INFRASTRUCTURE

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#### Learning Objectives

- Understand: -
  - Where are the Markets
  - Where hydrogen is used
  - Current progress and ideas

## Where would Hydrogen be used?

@####...don't say Fuel Cells...####@

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#### Contents

- 1. Current Hydrogen Applications and Uses
- Hydrogen Markets and Costs of Hydrogen
- 3. Distribution and Infrastructure
- 4. Hydrogen as a Vehicle Fuel
- 5. Hydrogen as a energy store
- 6. The future

## 1. Hydrogen Applications

#### Hydrogen Applications

- Ammonia synthesis
- Crude oil refining
- Methanol production
- Hydrogenation of fats in food processing
- Metallurgy
- Cooling in gas turbines
- Production of artificial diamonds
- Chemical industry
- Energetic use
- Town gas

#### Hydrogen Applications

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#### Ammonia Synthesis

- Process:
  - $\circ$   $N_2 + 3H_2$  ---->  $2 NH_3$  (Haber-Bosch process)
- Usage:
  - Cooling agent (ammonia refrigeration)
- Basis for products:
  - Fertilisers, explosives, fibers/plastics, pharmaceuticals, pulp & paper
- H<sub>2</sub> annual demand
  - 33 x 10<sup>9</sup> Nm³ in EU (55% of total)
  - 250 x 10<sup>9</sup> Nm³ worldwide
- 50% of total hydrogen consumption (was: 80% in the 1980ies)



#### Crude Oil Refining 1/2

- Multi-stage Process:
  - Distillation Tower at atmospheric pressure:
    - o division into 4 fractions:
      - kerosene, light/heavy gas oil, fuel oil (,Naphtha')
  - Cracking by heating, catalytic reforming and fluid catalytic reforming follow for various qualities of fuel
  - Alternatively "Hydrocracking" with hydrogen at ca. 150 bar also delivers various fuel stock under flexible production conditions
- Reforming of naphtha leads to excess hydrogen production from paraffin cracking.
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## Crude Oil Refining 2/2

- Hydrogen usage:
  - de-sulphurisation of intermediate products (to prevent poisoning of catalysts and/or improve quality) by forming H<sub>2</sub>S
  - Hydrocracking
- Production by
  - naphtha reforming (by-product)
  - fuel gas/methane reforming
- H<sub>2</sub> annual demand
  - 17 x 109 Nm³ in EU (30% of total)
  - 185 x 109 Nm³ worldwide
- 37% of total hydrogen consumption

#### Methanol Synthesis 1/2

- Two-stage Process:
  - Steam reforming of natural gas (Syngas, H<sub>2</sub> and CO)
  - Catalytic (metal oxide) formation of methanol under pressure and high temperature
- Followed by purification by distillation
- Alternative:
  - Synthesis from CO<sub>2</sub> and H<sub>2</sub>
- Energy demand is higher, though, but no fossil fuels necessary if hydrogen is available from other sources



#### Methanol Synthesis 2/2

- Products:
  - Methanol
  - MTBE (methyl tertiary butyl ether, gasoline additive)
  - Formaldehyde
  - Others.
- H<sub>2</sub> annual demand
  - 4 x 109 Nm³ in EU (7% of total)
  - 40x 109 Nm³ worldwide
- 8% of total hydrogen consumption



#### Hythane

- "Synthetic" Town Gas made from natural gas and hydrogen
  - Possible interim energy carrier for transporting hydrogen via the natural gas network
  - Hydrogen content of 5 to 15% supposedly has no (or hardly any) influence on gas appliances
  - Hydrogen content of up to 30 to 50% would require changes to gas appliances, but no replacement of installations



## Further Applications 1/2

- Food processing: hydrogenation of plant fats for hardening (margarine and cooking fat), inert gas in processing
- Metallurgy: use of pure hydrogen/oxygen flames in cutting/welding/brazing, annealing, reduction of metal oxides
- Power industry: generator cooling
- Electronics industry: inert atmospheres for semiconductor producing furnaces

#### Further Applications 2/2

- Glass industry: high purity flames for cutting and welding, artificial gemstone/diamond production, high purity atmospheres in furnaces
- Space industry: rocket fuel
- Balloon gas
- Chemical industry: styrene, ethylene, peroxide production

#### Future Applications

- Natural gas substitute:
  - Gas heaters/boilers, cookers, motors
- Electricity storage medium:
  - Electrolytic conversion to electricity and back to electricity by fuel cell
- Fuel cells:
  - Emission free (locally) cogeneration of heat and electricity
- Hydrogen vehicles:
  - Emission free (locally) vehicles with fuel cells or hydrogen
     ICE (internal combustion engine)



#### 2. Hydrogen Markets

#### Market History

- Hydrogen was first used at large scale with the balloons and airships coming into use at the end of the 19th century - leading up to the Zeppelin development
- Many of the now well-known material problems were then first encountered (steel embrittlement, requirements for sealings and washers/gaskets, leakage rates from storage vessels etc.)
- Large scale employment of hydrogen was then spurred by the fertiliser industry in the early 20th century

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#### Market History

- Town gas in the UK till 1950's
- Up to the mid-nineties many town gas networks in Germany (esp. East Germany and Berlin), in 1992 3 x 1012 Nm³ (being approx. 10% of the pipeline supplied gas in German energy supply)
- Town gas supply still in operation in Stockholm City.

#### Size of the Market

- Hydrogen production
  - ca. 500 x 10<sup>9</sup> Nm³ per year worldwide
  - ca. 60 x 10<sup>9</sup> Nm<sup>3</sup> per year in EU (ca. 180 GWh)
  - Annual rise since year 2000 ca. 5 to 10%
- Comparison:
  - world natural gas market 2.4 x 10<sup>12</sup> Nm³ p.a.
  - EU natural gas market 470 x 10<sup>9</sup> Nm<sup>3</sup> p.a. (ca. 5 TWh)

#### Market Segments

• EU statistics:

Captive

By-product

Merchant

Direct energetic

Surplus (stranded)

ca. 55 x 10<sup>9</sup> Nm<sup>3</sup> (93%)

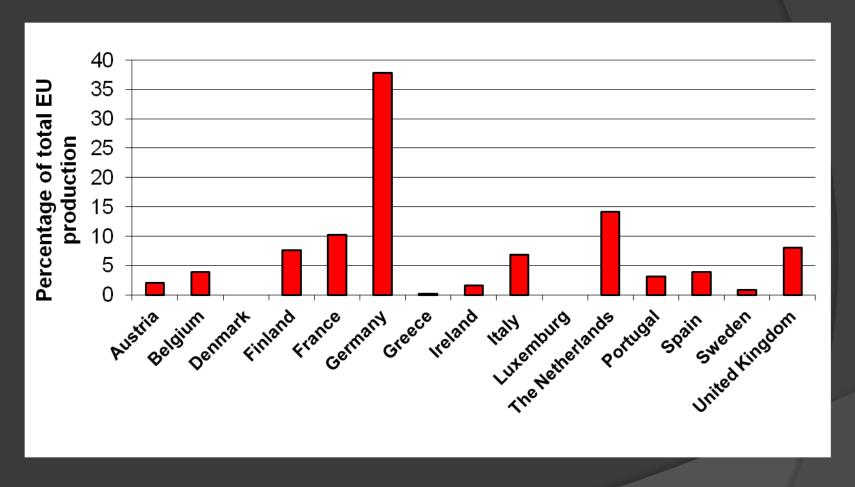
ca. 3 x 10<sup>9</sup> Nm<sup>3</sup> (5%)

ca. 1.2 x 10<sup>9</sup> Nm<sup>3</sup> (2%)

ca. 0.5 x 10<sup>9</sup> Nm<sup>3</sup>

ca. 0.015 x 10<sup>9</sup> Nm<sup>3</sup>

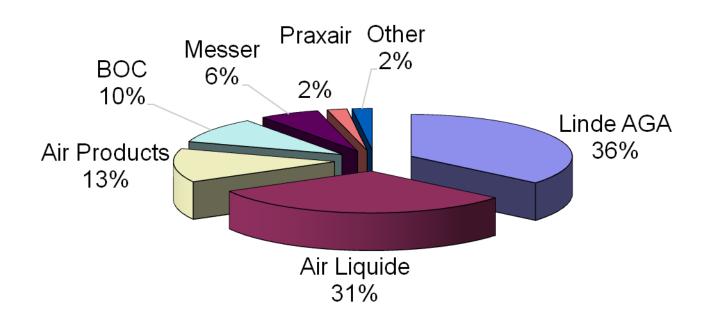
#### **European Production**



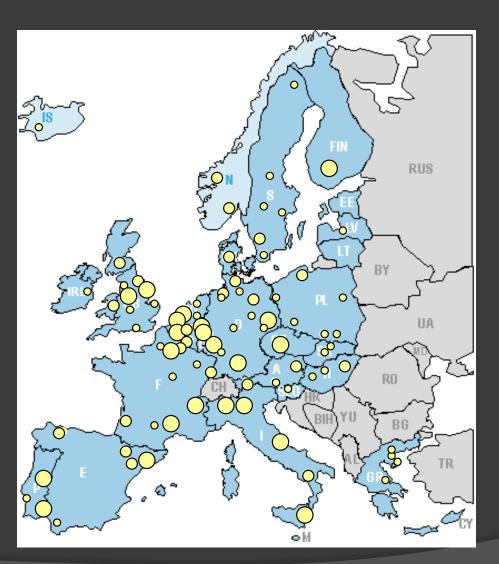
- EU figures for ca. 1997
- Out of total of 60 x 10<sup>9</sup> Nm<sup>3</sup> p.a

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#### Key Players



#### Where in the EU?



- \* Production Site (1 to 3)
- Production Site (3 to 7)
- Production Site (> 7)

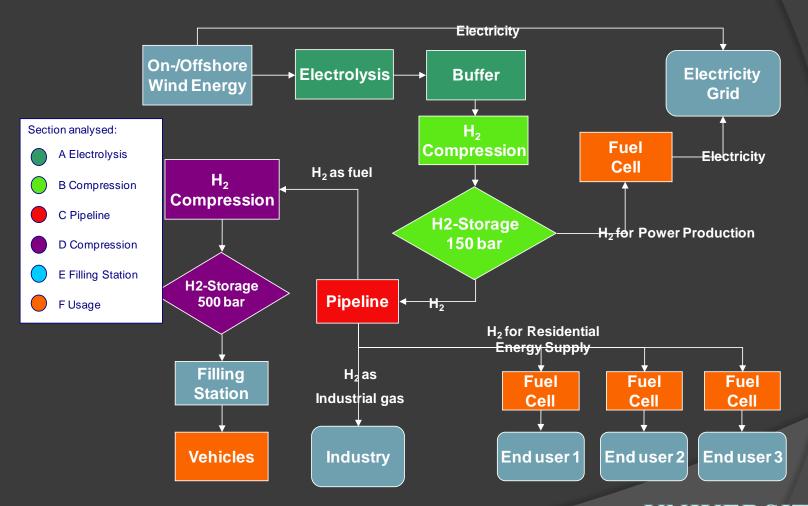
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## Hydrogen Costs

Hydrogen Quality	3.0	3.0	5.0	5.0
Item	Costs [EUR]	Costs [EUR]	Costs [EUR]	Costs [EUR]
10 l bottle / 1.8 Nm <sup>3</sup>	48.10		61.50	
50 l bottle / 8.9 Nm <sup>3</sup>		67.80		111.90
Energy Surcharge per Bottle	1.00			
Surcharge for Transporting Hazardous Materials	18.00			
Rental Fee per Bottle / daily	0.39			
Rental Fee per Bottle / flat rate for 1.5 years	160.00			
Effective Costs per Nm <sup>3</sup> (excluding bottle rental)	40.00	9.71	46.00	16.74

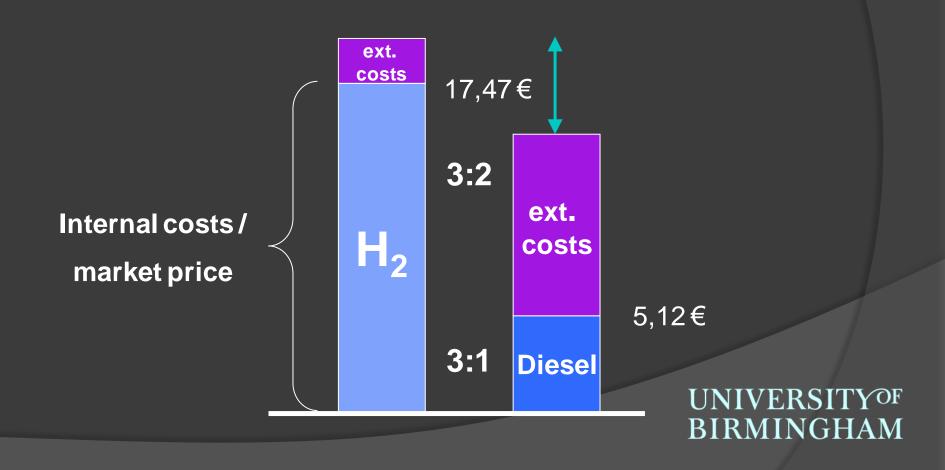
#### Hydrogen Costs



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#### Cost Breakdown

3: 1 Ratio Exists on comparison.....



#### Why the difference in costs?

#### Internal costs:

- costs of production and delivery of vehicles and fuel
- taxes and levys
- market price (as a sum of the above plus company profit)

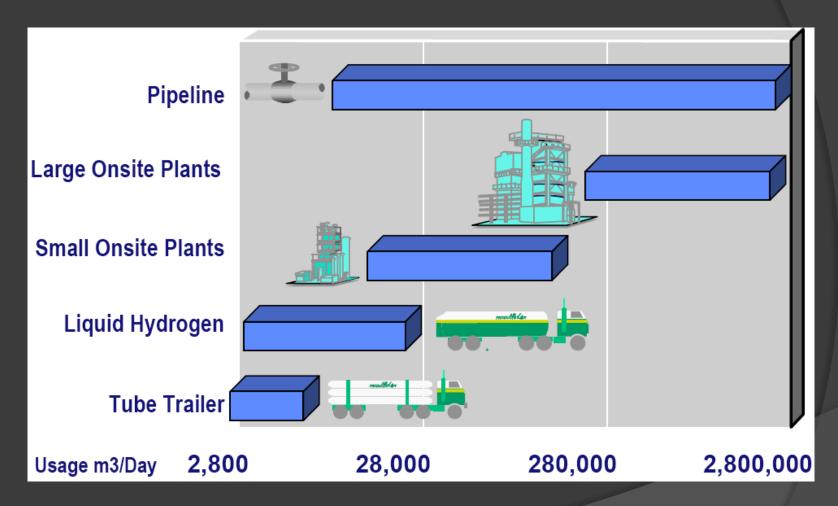
#### External costs

- costs caused for the society, but not attributed or attributable to single products or services
- health services due to environmental pollution
- health and other services due to noise pollution
- public services in safety, accident prevention etc.
- general costs of land use, rain run-off management etc.

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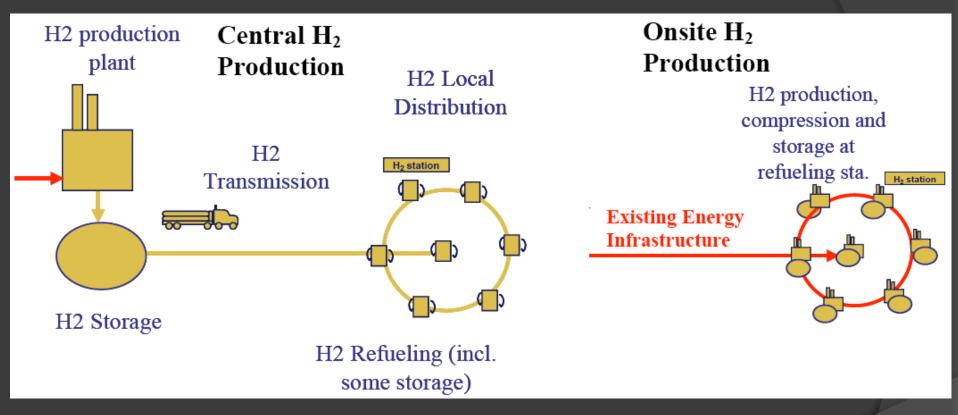
#### 3. Hydrogen Distribution

#### Distribution



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#### The process



Joan Ogden and Christopher Yang, "Implementing a Hydrogen Energy Infrastructure: Storage Options and System Design" (UCD-ITS-RR-05-28), 2005

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#### Distribution

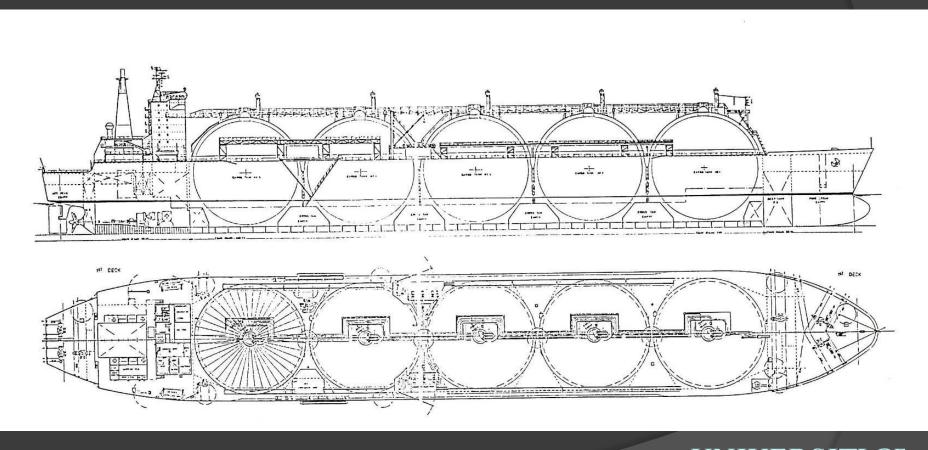
- Conventionally distributed
  - Gas Cylinders / Large Vessels
  - Liquefied tanker
  - Liquefied pipeline
- Large scale pipeline?
  - Expensive
- Existing lines
  - Dilute hydrogen to 10%
  - Changeover problems?



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Photographs courtesy of PLANET and HyNet

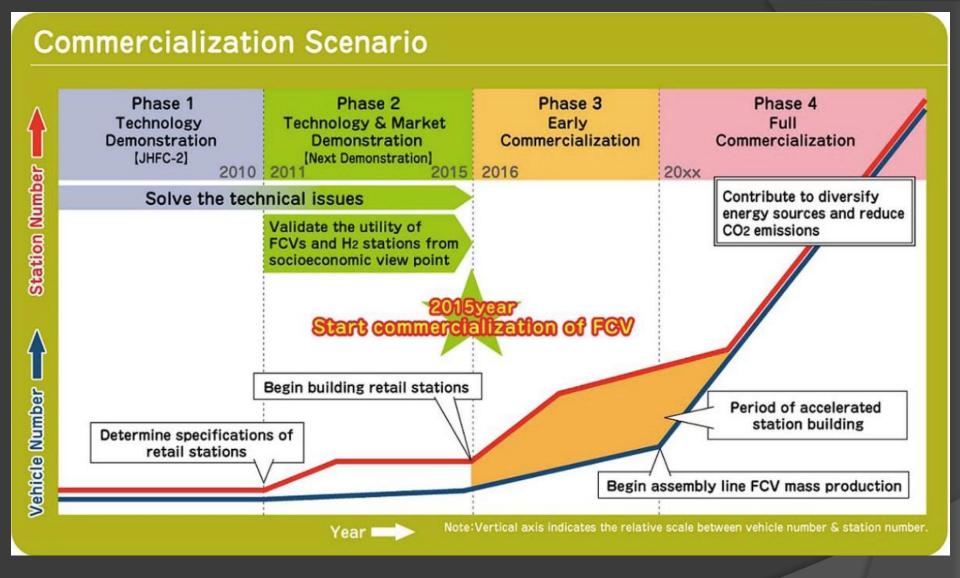
# Marine Transportation: Gas/Liquid Tanker



#### Pipeline



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Leading Japanese automakers & energy companies have agreed on a scenario which sees commercialization of FCVs and hydrogen stations beginning in 2015.

FCCJ, http://www.fccj.jp/pdf/20080704sks1e.pdf

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# 4. Hydrogen in the vehicle Fuel Market: Refuelling infrastructure

### ...Re fuelling

- Pressure drop between fuelling station and vehicle should be around 150 bar for fuelling process < 10 minutes
- From vehicle tank end-pressure 350 bar it follows that the storage pressure needs to be around 500 bar
- Due to the negative Joule-Thompson effect hydrogen will heat up when filling vehicle tank; subsequent cooling to ambient temperature reduces pressure in tank -> overfilling' necessary
- at vehicle tank pressure 700 bar, storage pressure of ca.
   1.000 bar would be necessary -> other filling technologies than pressure drop, for instance cooling during filling (isothermal filling)

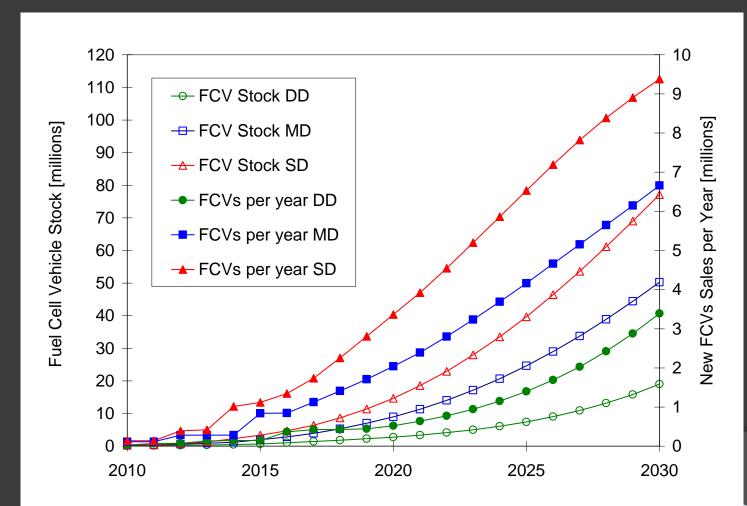
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## Need filling stations

- Number of (conventional) fuelling stations in Germany:
   15,000
- Expense to convert a fuelling station to hydrogen:
  - ca. € 1M
- Number of gasoline filling stations in U.S. ca. 170.000

Gasoline (LNG) distribution solely through road transport

#### US Scenario - Vehicles

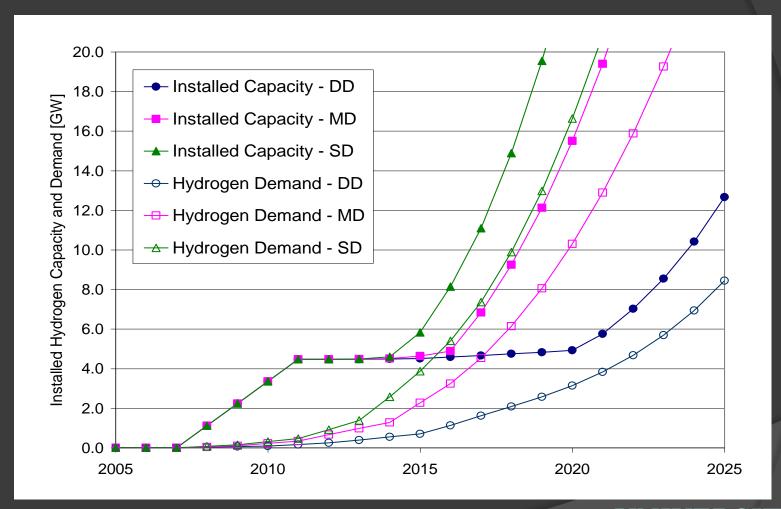


**Projection for U.S.:** 

- DD delayed
- MD moderate
- SD successful development

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### US Scenario - Stations

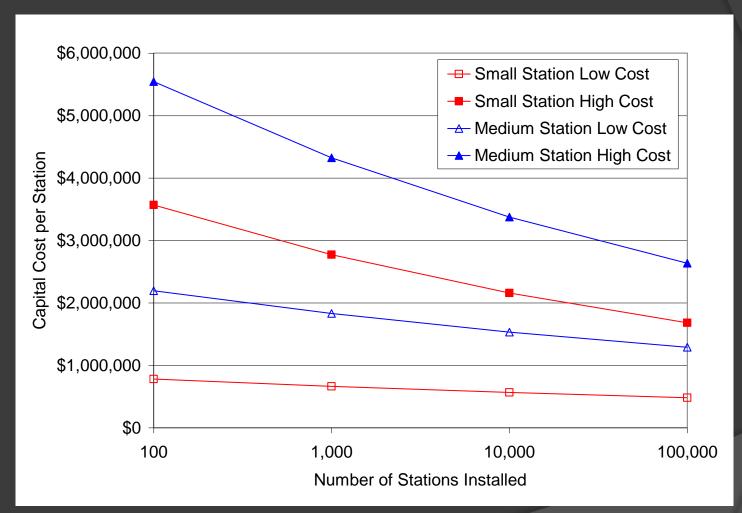


#### **Projection for U.S.:**

- DD delayed
- MD moderate
- SD successful development

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#### Costs of Fuel Stations



Projection for the U.S.

- small = 500 kg  $H_2$  p.day
- medium =  $2.500 \text{ kg H}_2 \text{ p.day}$

# Filling Stations

Cute Project





# Components







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# Cute Project Stations



# Liquid Hydrogen Stations



### **SMR Station**



# 5. Hydrogen as a renewable energy store

# Why do we need energy stores?

- Reduction of fluctuations
- Temporal shift of power
- Integration of surplus/stranded production

# Renewable energy sources

Source	Туре	Heat	Electricity	Other
Solar	thermal	Emile Emile		
	PV		Emile Marie Marie	
	tower		Em Em	
	algae	( <b>E</b> )	( \( \xi_{\text{M}} \)	H <sub>2</sub>
Wind	repellor		Emile Emile	
Solar/Wind	thermal flow		Emile Emile	
Geothermal			Emile Sund	
Hydro	stream		Emile Emile	
	dam		Em Em	
Tidal/ wave			Emile Emile	
Biomass	combustion		Emile Emile	
	fermentation	( <b>E</b>	( <b>E</b> M <b>E</b> )	CH <sub>4</sub>
	pyrolysis	( Em )	( Emit )	syn-gas

#### Storage Methods

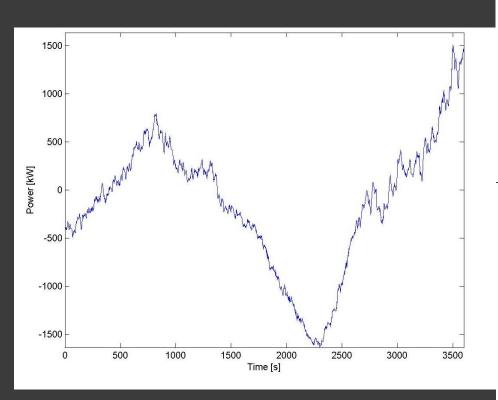
#### Conventional

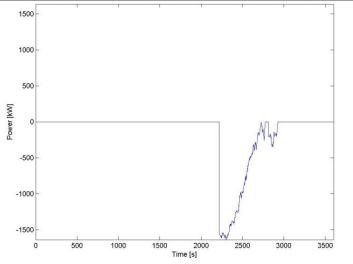
- pumped storage
- Batteries
- flywheels
- Novel
  - compressed air
  - super-caps
  - hydrogen

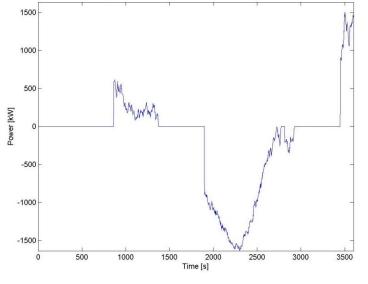
- limited by topography
- capacity tied to device itself
- limited capacity

- limited efficiency
- limited efficiency
- arbitrary storage size

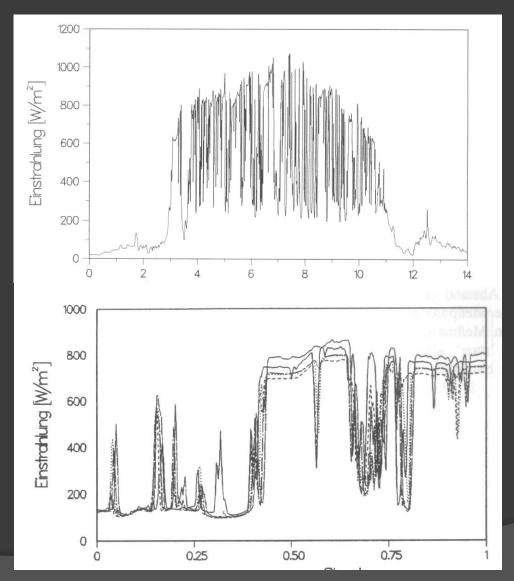
# Typical energy use





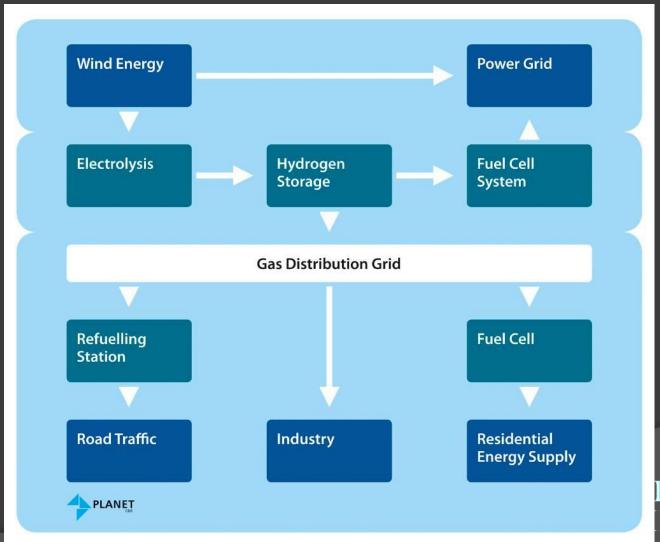


# **Typical Energy Production**



Solar production

# Why is hydrogen important?



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# What couplings?

- Sources (realistic):
  - Wind energy (excess and sinking feed in tariffs)
  - Biomass (gasification, reforming of biogas/methane)
  - Hydro
- Far away:
  - Photovoltaics
  - Thermolysis with solar energy



### 6. The Future

#### **UK Investment**







BAXI CHP H<sub>2</sub> Filling Station 5 x Hybrid Vehicles

# House in the West Midlands heated by a Hydrogen Fuel Cell

**PEMFC** 

Fuel: Natural Gas

The fuel cell has a dual purpose:

- (i) Supply of electricity
- (ii) Heating





### CABLED



□ £25 national project (£7million West Midlands)



Hydrogen Refueling Station UoB



Royal Mail Hydrogen ICE and HFCHV
UoB

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# PRESENT Hydrogen LOCATIONS



# NEW 2013 LOCATIONS



Bergen

# Next year.....



# Summary

- Hydrogen is versatile
- Used in many different and broad applications

- Distribution is difficult at present
  - Projects are on going to improve