

# Hydrogen Fires

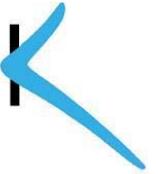
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# Outline (Lecture 1)



- **Basic characteristics of hydrogen fires**
- **Some fundamentals of fires**
- **Hydrogen jet fires**
  - Flame Length
  - Lift-off heights
  - Flame radiation
  - Horizontally oriented high pressure hydrogen jet flames
  - Ignition of Hydrogen Leaks

# Basic characteristics of hydrogen fires

- Hydrogen fire burns quickly
- Hydrogen flames have low radiant heat
- Hydrogen fires are almost invisible



# Basic characteristics of hydrogen fires

Table 3 Combustion time for 121 litres of fuel (reproduced from [1])

Fuel	Combustion time
Liquid hydrogen (LH <sub>2</sub> )	27 sec
Propane	4 min
Petrol	5 min
Jet fuel (JP-4)	7 min



Comparison of fire from a leak in a gasoline engine car and the same kind of leak from a hydrogen car. Gasoline car to the right and hydrogen car to the left (reproduced from [2])

**Table A7.1**  
**Hydrogen Accidents - Industrial**

Category	Number of Incidents	Percentage Total Accidents
Undetected <b>Leaks</b>	32	<b>22</b>
Hydrogen-Oxygen Off-gassing Explosions	25	17
Piping and Pressure Vessel <b>Ruptures</b>	21	<b>14</b>
Inadequate Inert Gas Purging	12	8
Vent and Exhaust System Incidents	10	7
Hydrogen-Chlorine Incidents	10	7
Others	<u>35</u>	<u>25</u>
<b>Total</b>	<b>145</b>	<b>100</b>

Zalosh, R. G., and T. P. Short. *Comparative Analysis of Hydrogen Fires and Explosion Incidents*. C00-4442-2, Factory Mutual Research Corp., Norwood, MA (1978).

Zalosh, R. G., and T. P. Short. *Compilation and Analysis of Hydrogen Accident Reports*. C00-4442-4, Factory Mutual Research Corp., Norwood, MA (1978).

# Hydrogen Fire Hazards

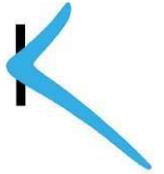
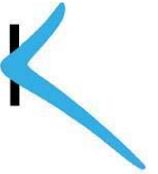


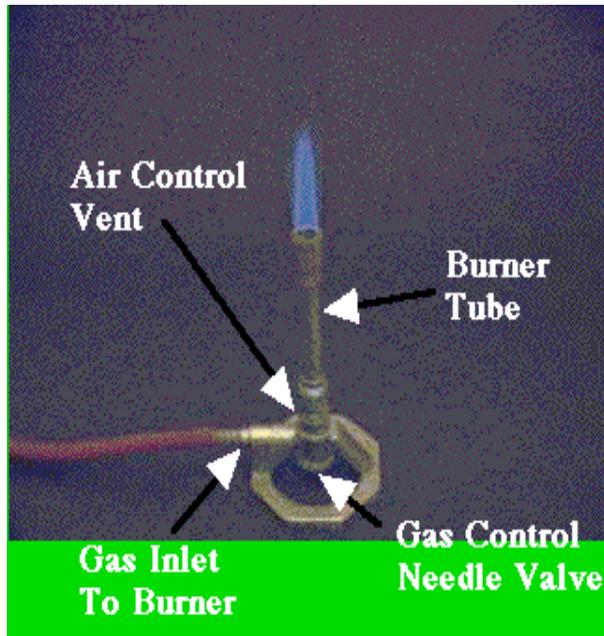
Table 11 Frequency of occurrence of ignition sources (reproduced from [47])

Ignition source	Hydrogen incidents		Non-hydrogen incidents	
	Number	%	Number	%
Arson	0	0	37	2.6
Collison	2	2.5	121	8.4
Flame	3	3.7	113	7.9
Hot surface	2	2.5	56	3.9
Electric	2	2.5	114	7.9
Friction spark	2	2.5	33	2.3
Not identified	70	86.3	942	64.5
Non-ignition	0	0	21	1.5
Total	81	100	1437	100

# Some fundamentals of fires



- Premixed combustion



- Non-premixed combustion



Partially premixed combustion

# Fireballs, jet, pool and flash fires

- A fireball results from the burning of fuel-air cloud
- Jet fires result from the combustion of a fuel released from a pressurized system or storage vessel.
- Pool fires are the result of surface burning of flammable or combustible liquid
- A flash fire is the non-explosive combustion of a vapour cloud resulting from a release of flammable or combustible material into the open air.

# Jet Fires

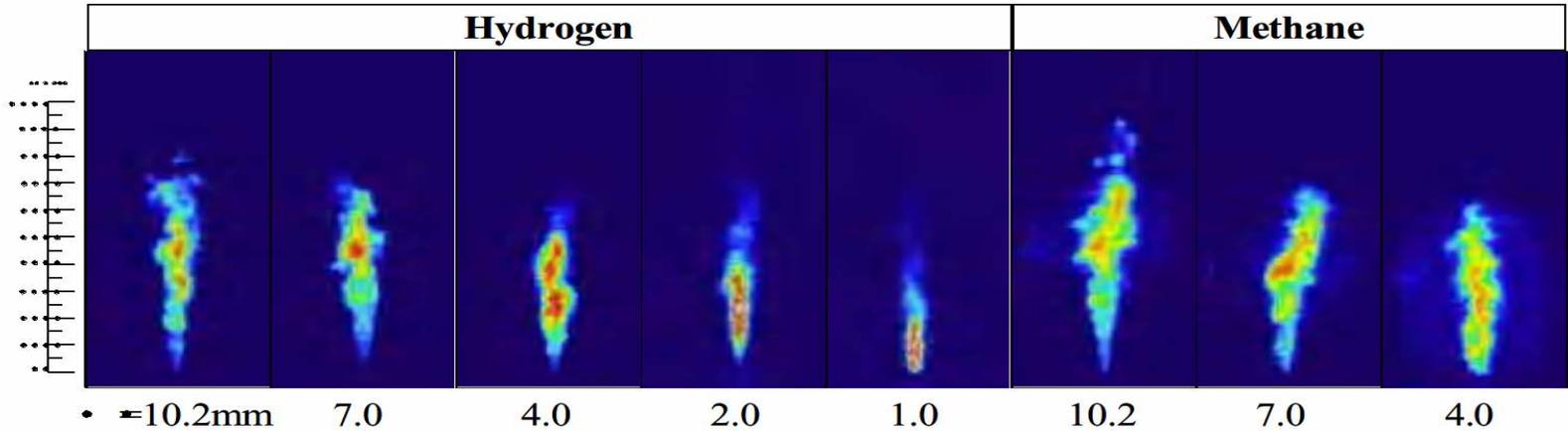
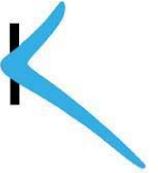
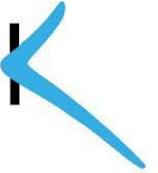


Figure 6 The infrared pictures of hydrogen and methane flames in relation to gas nozzle diameter.

Table 1 Comparison of hydrogen and methane flame

		Hydrogen	Methane
Upward Release	Maximum Flame Temperature	1,171 • •	928 • •
	Nozzle Dia. $\Phi$ 4.0mm		
	Flame Size	Length: 710mm	Length: 830mm
	Nozzle Dia. $\Phi$ 10.2mm	Width: 120mm	Width: 150mm
	Maximum Heat Release	1.6kW/m <sup>2</sup>	1.8kW/m <sup>2</sup>

# Fireball



**Liquid Hydrogen and Oxygen Fireball  
230 kg (500 lb) Test**

# Hydrogen jet fires - Flame Length

- **Delichatsios's correlation**

- Nondimensional Froude number

$$Fr_f = \frac{u_e f_s^{3/2}}{(\rho_e / \rho_\infty)^{1/4} \left[ \frac{\Delta T_f}{T_\infty} g d_j \right]^{1/2}}$$

- Nondimensional flame length

$$L^* = \frac{L f_s}{d_j (\rho_e / \rho_\infty)^{1/2}} = \frac{L f_s}{d^*}$$

- In the buoyancy-dominated regime,

$$L^* = \frac{13.5 Fr_f^{2/5}}{(1 + 0.07 Fr_f^2)^{1/5}}$$

for  $Fr_f < 5$

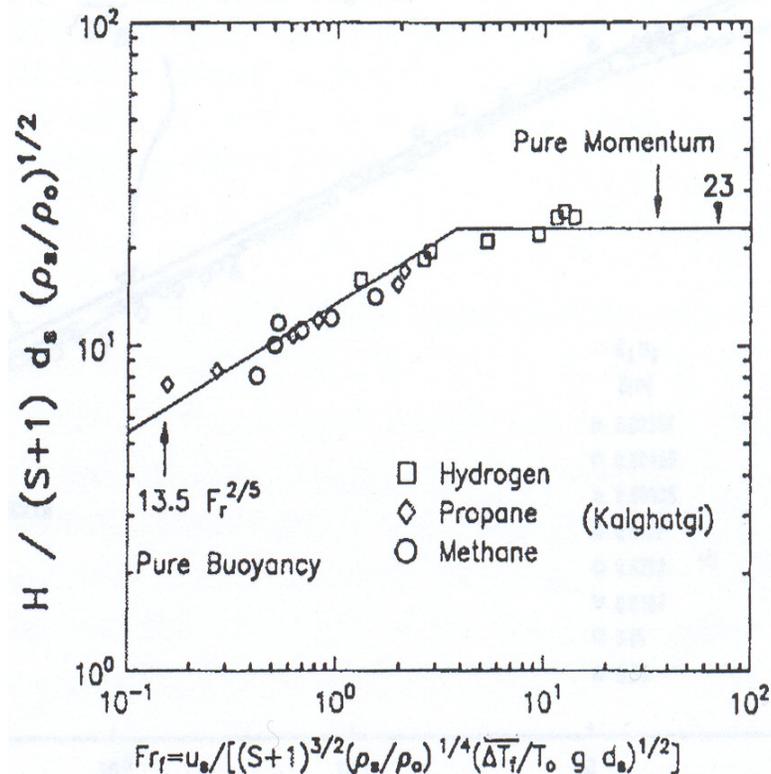
- In the momentum-dominated regime,

$$L^* = 23$$

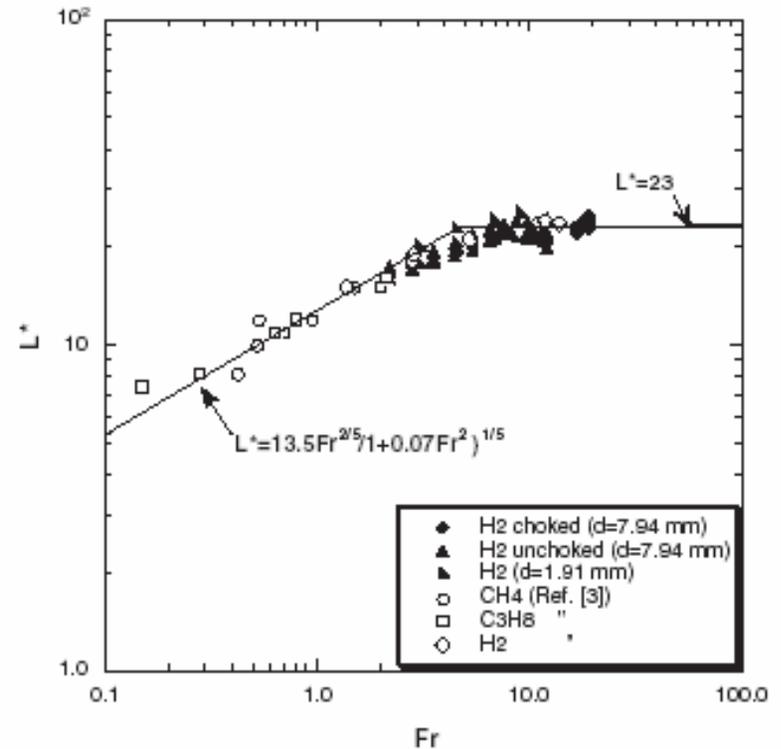
for  $Fr_f < 5$

# Hydrogen jet fires - Flame Length

- Delichatsios' correlation

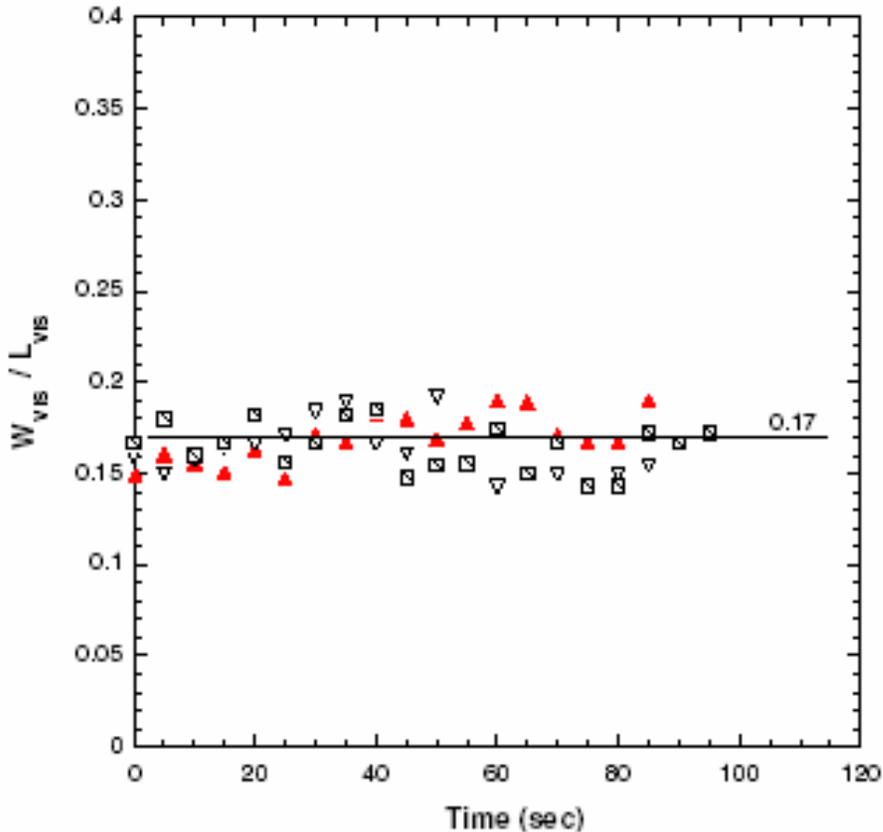


- With Sandia vertical hydrogen flame data



# Hydrogen jet fires - Flame Width

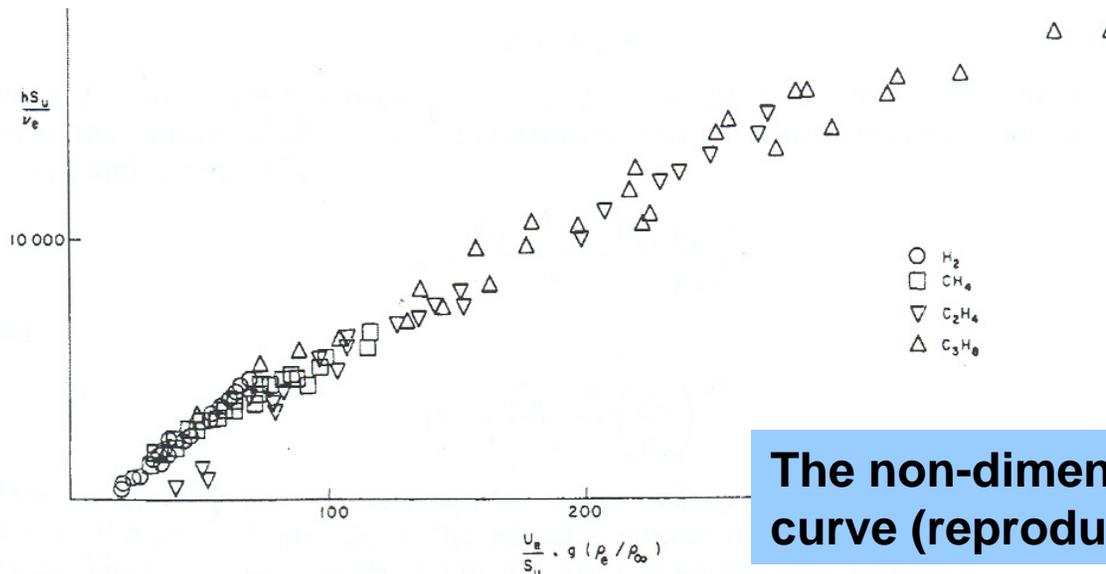
Variation of visible flame width with Froude number. Data Ratio of maximum flame width to flame length determined from visible flame emission (reproduced from [20]).



# Hydrogen jet fires – Lift-off height

- Kalghatgi's correlation from subsonic jets to those from under-expanded sonic jets:

$$\left( \frac{h \rho_e S_{L \max}}{\mu_e} \right) = C \left( \frac{U_e}{S_{L \max}} \right) \left( \frac{\rho_e}{\rho_\infty} \right)^{1.5}$$



The non-dimensional lift-off height curve (reproduced from [12])

# Radiation from jet flame

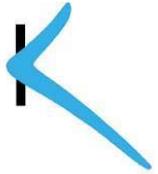
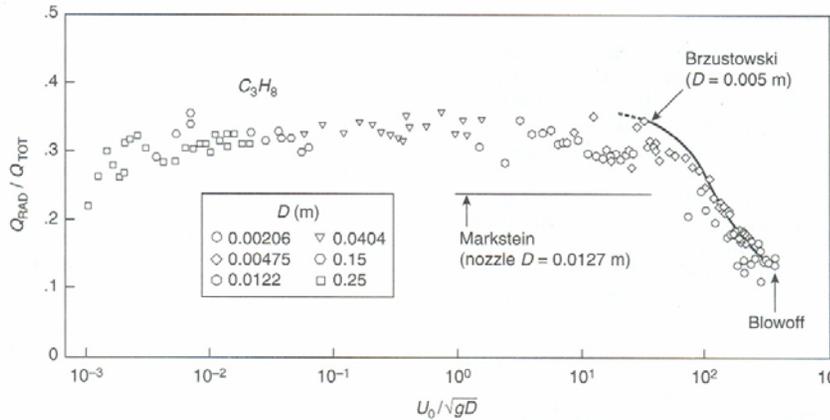
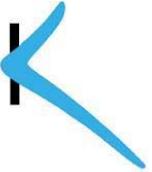


Table 7 Comparison of radiative fraction,  $\chi_r$ , of various fuels (reproduced from [30])

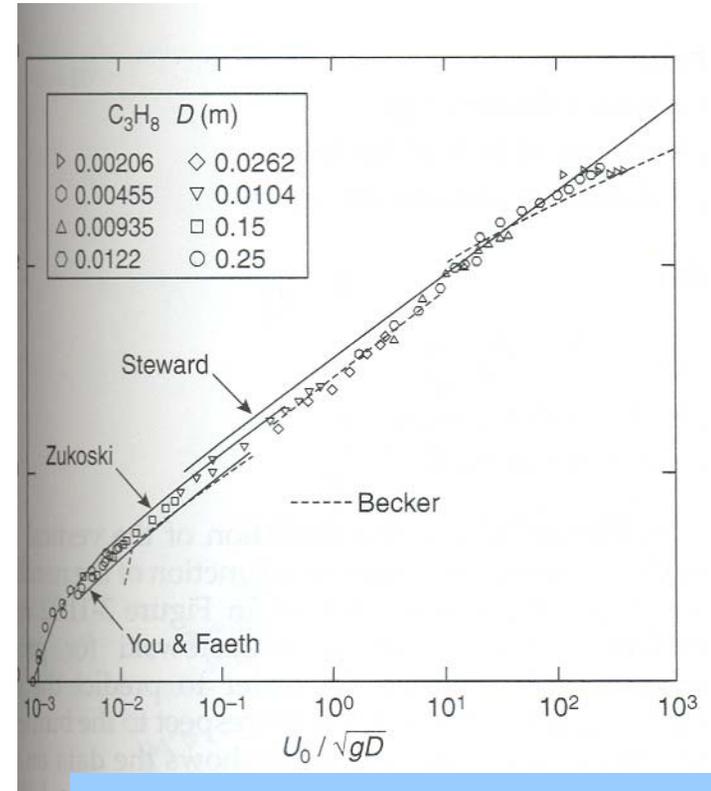
Fuel	$f_s$	$\chi_r$ Brzustowski [34]	$\chi_r$ Burgess and Hertzberg [33]	$\chi_r$ Tan [35]	$\chi_r$ Kent [36]	$\chi_r$ McCaffrey [33]
Hydrogen	1.0	0.2	0.17	-	-	-
Methane (C <sub>1</sub> )	0.189	0.2	0.23	0.20	0.19	0.22
Ethylene (C <sub>2</sub> )	0.170	0.25	0.36	0.26	0.25	0.38
Propane (C <sub>3</sub> )	0.176	0.30	-	0.32	0.32	0.302
Butane (C <sub>4</sub> )	0.175	0.30	0.30	0.37	0.37	-
C <sub>5</sub> and higher	-	0.40	-	-	-	-

# Radiation from jet flame



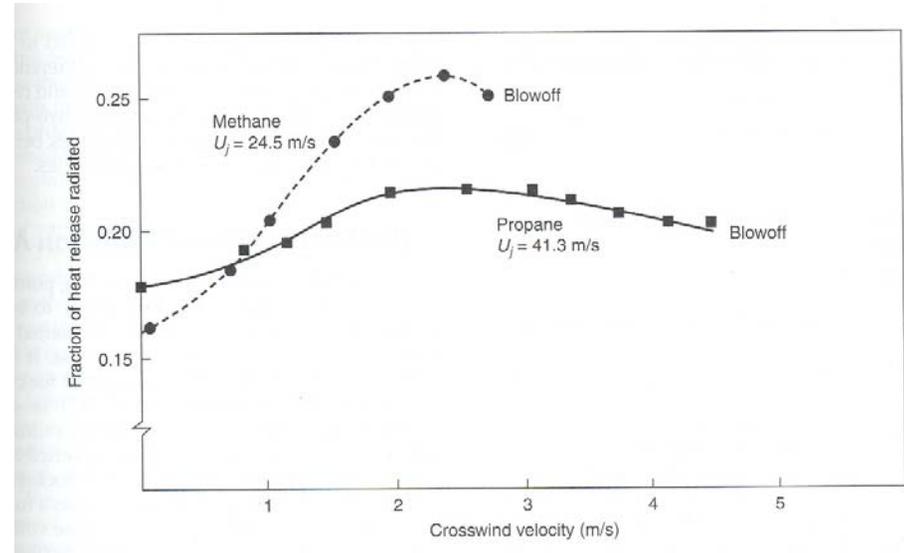
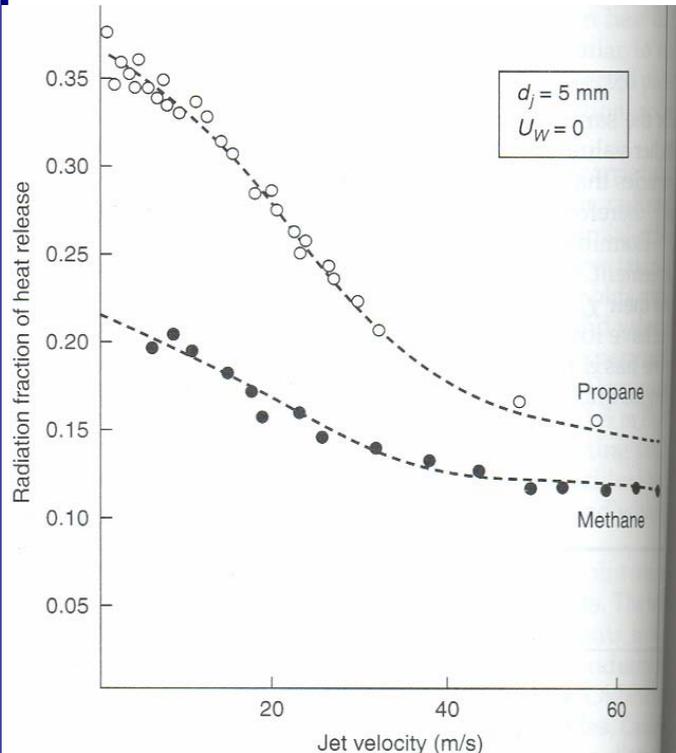
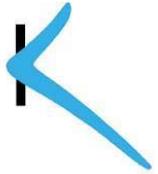
Radiative fraction measured by McCaffrey [33].

Radiative fraction is constant in the buoyancy-controlled regime, but for momentum-controlled jet flames, the radiative fraction decreases until blow-off occurs.



Flame height per nozzle diameter as a function of (reproduced from [30]).

# Radiation from jet flames



Effect of crosswind on the radiant energy

Radiation fraction  $\chi_r$  decreases with increasing jet velocity

# Radiative properties of hydrogen jet fires

- (Schefer et al. [20], Molina and Schefer [21] and Schefer et al. [22])

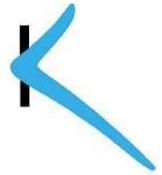


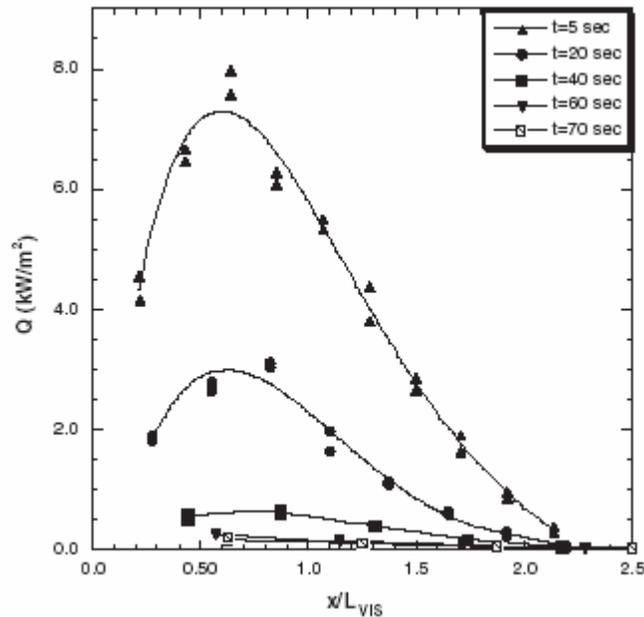
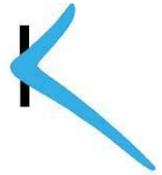
Table 8 Flame conditions (reproduced from [20])

Flame	$d_J$ (mm)	$u_J$ (m/s)	$Q$ (slm)	$Re$	$m$ (g/s)	$m\Delta h_c^a$ (kW)
Lab flame	1.91	87.7	15	1569	0.021	2.64
	“	116.3	20	2081	0.028	3.34
	“	174.5	30	3123	0.042	5.01
	“	261.6	45	4686	0.062	7.52
	“	349.0	60	6247	0.083	10.0
SRI flame (s)	7.94					
$t = 5$	“	1233	41,026	$9.8 \times 10^5$	57.3	6874
$t = 20$	“	1231	16,589	$3.9 \times 10^5$	23.17	2779
$t = 40$	“	1078	4954	$1.2 \times 10^5$	6.92	830
$t = 60$	“	644	1482	$4.2 \times 10^5$	2.07	248
$t = 70$	“	446	810	$1.9 \times 10^5$	1.13	135

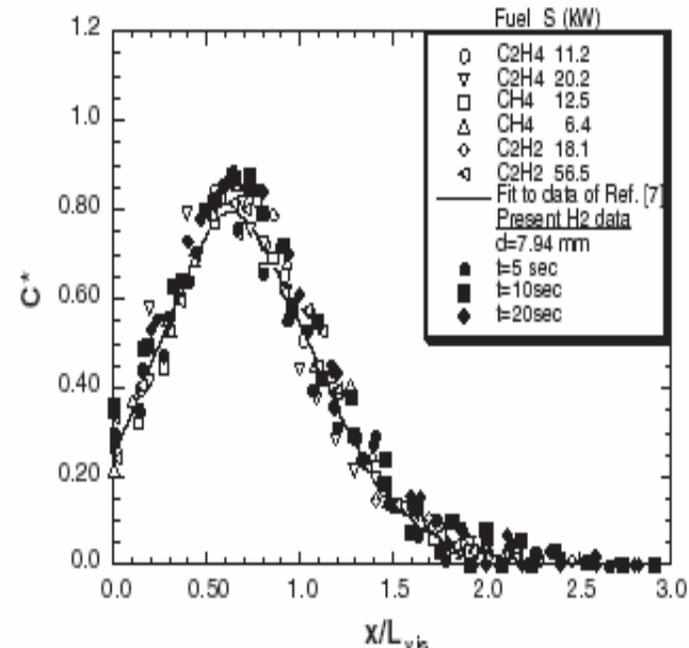
<sup>a</sup> $\Delta h_c = 119,962$  kJ/kg for hydrogen heat of combustion.  $\tau = (\rho_f W_f^2 L_{vis} f_s) / (3\rho_0 d^2 u_J)$  [11].

# Radiative properties of hydrogen jet fires

- (Schefer et al. [20], Molina and Schefer [21] and Schefer et al. [22])

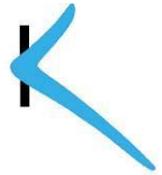


Profiles of radiative heat flux along the centreline of a turbulent, hydrogen-jet flame. Jet diameter is 7.94 mm. Jet orientation is vertical (reproduced from [20]).



Profiles of normalized radiative heat flux along the centreline of a turbulent, hydrogen-jet flame. Jet diameter is 7.94 mm. Jet orientation is vertical (reproduced from [20]).

# Radiative properties of hydrogen jet fires



The radiative fraction for H2 flames is nearly a factor of two lower than non-sooting hydrocarbon flames

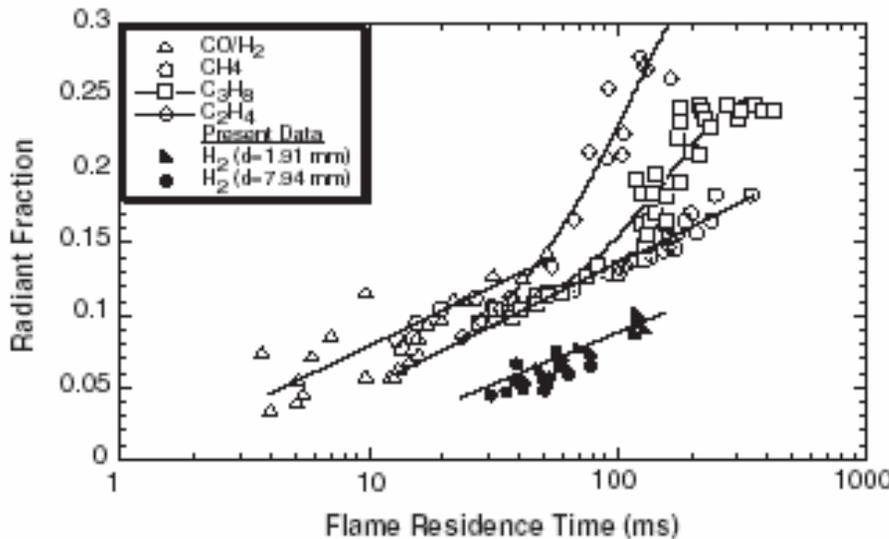


Fig. 24 Radiant fraction as a function of flame residence time (reproduced from [20]).

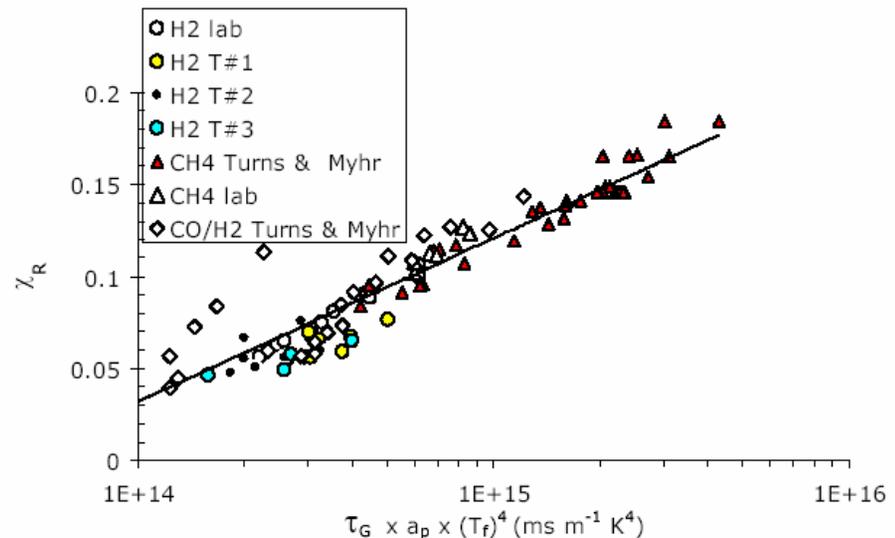
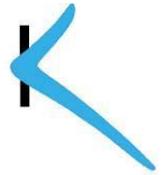


Fig. 25 Variation of radiative fraction ( $\chi_{rad}$ ) with the factor  $\tau_G \times a_p \times T_{ad}^4$  for the gas mixtures in Fig. 24 (reproduced from [21]).

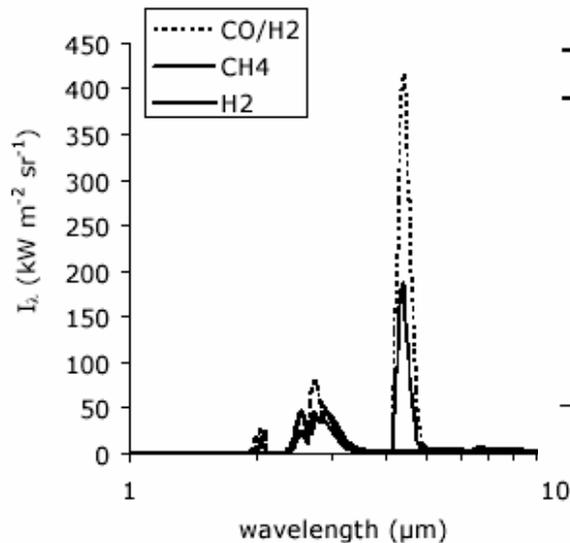
# Radiative properties of hydrogen jet fires

- (Schefer et al. [20], Molina and Schefer [21] and Schefer et al. [22])



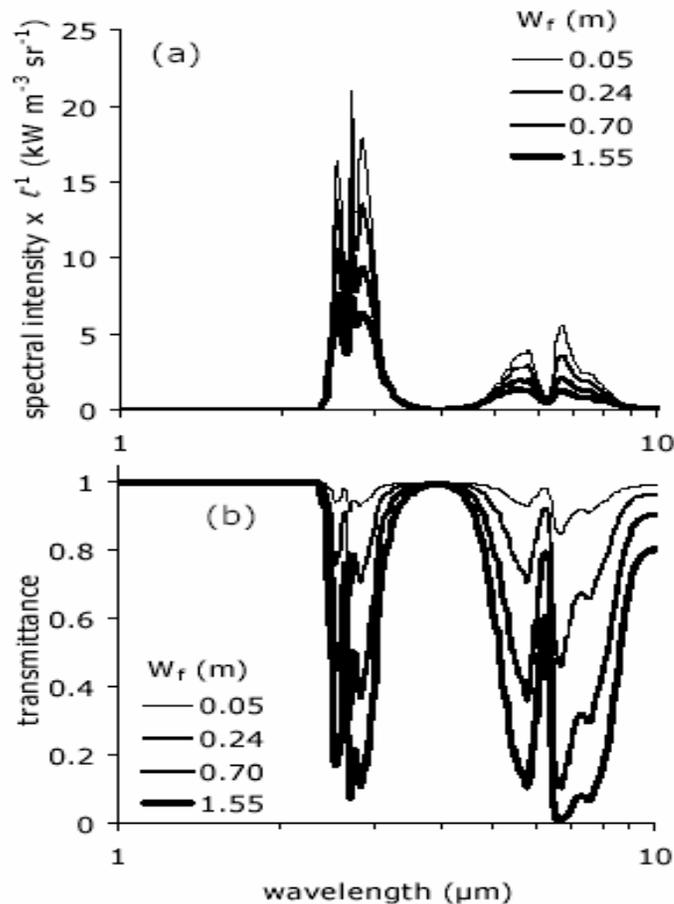
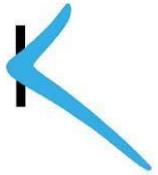
Radiative fraction of large-scale jet flames was smaller than that predicted by the correlation obtained for laboratory-scale flames.

Table 10 Flame width and optical thickness ( $\kappa = a_p \times W_f/2$ ) for the different gas mixtures (reproduced from [21])

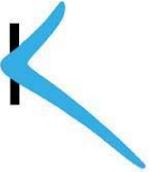


Gas mixture	$W_f$ (m)	$\kappa$ (-)
H <sub>2</sub> lab [17]	0.059 - 0.085	0.01 - 0.02
H <sub>2</sub> T#1 [16]	0.363 - 1.555	0.04 - 0.18
H <sub>2</sub> T#2 [16]	0.246 - 0.705	0.03 - 0.08
H <sub>2</sub> T#3 [17]	0.374 - 0.806	0.04 - 0.09
CH <sub>4</sub> lab [17]	0.086 - 0.098	0.02
Turns CH <sub>4</sub> [14, 27]	0.114 - 0.255	0.03 - 0.07
Turns CO/H <sub>2</sub> [14, 27]	0.030 - 0.098	0.01 - 0.04

# Radiative properties of hydrogen jet fires - (Schefer et al. [20], Molina and Schefer [21] and Schefer et al. [22])



**Fig. 27** Line-of-sight calculations (RADCAL [40]) of (a) spectral intensity divided by pathlength ( $l$ ) and (b) transmittance for flame A of the TNF Sandia database [43-45] ( $W_f = 0.05$  m) and for flames scaled with  $d_j$  to obtain different values of  $W_f$ . Predictions are for  $x/L_f = 0.7$ . (reproduced from [21])



# Horizontally oriented high pressure hydrogen jet flames

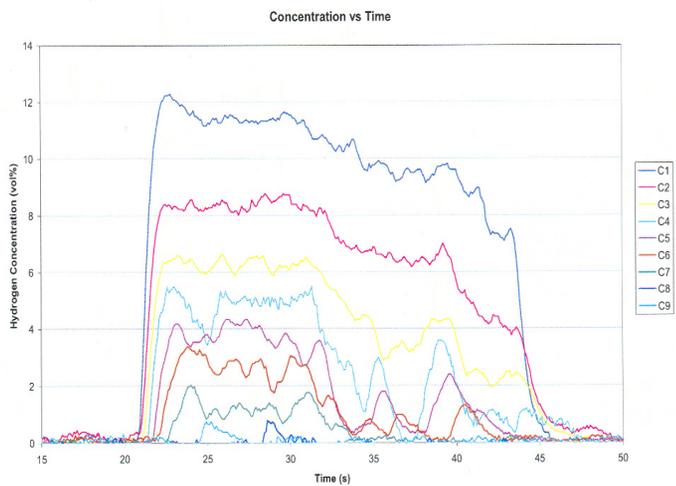


Fig. 29 Hydrogen concentrations at positions: 3, 4, 5, 6, 7, 8, 9 and 10 m downstream from the release point for the 135 barg release from a 3 mm orifice (reproduced from [38])

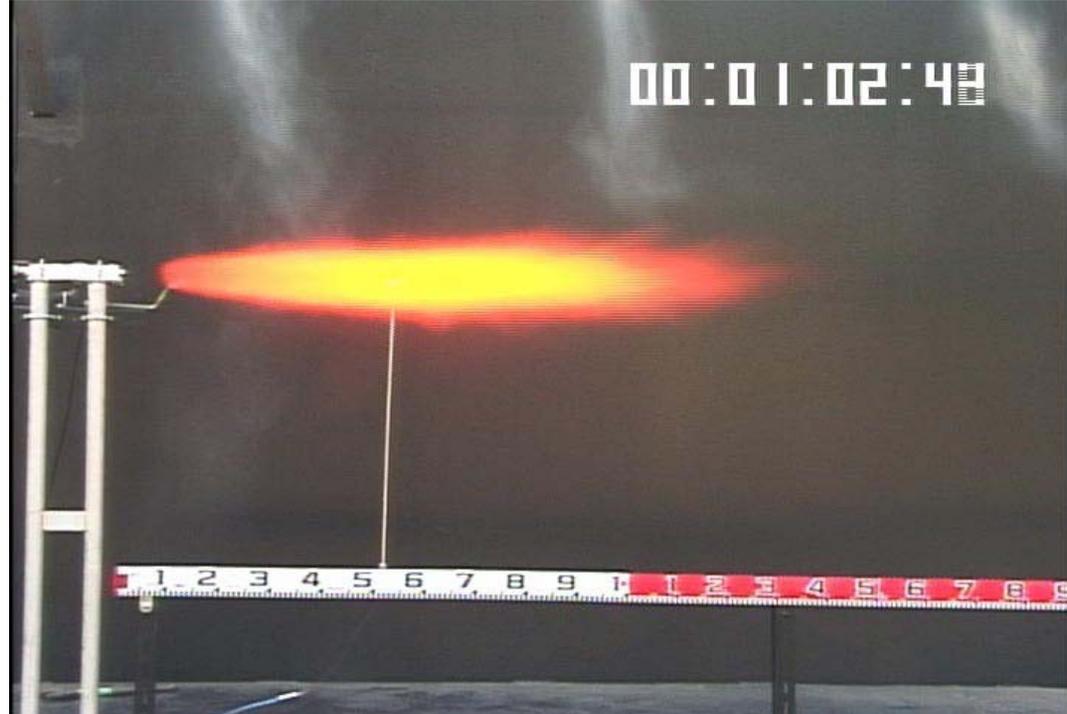
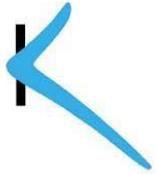


Fig. 31 Jet flame of 40 MPa high-pressure hydrogen (reproduced from [69]).

# Ignition of Hydrogen Leaks



- Astbury and Hawksworth [47]

Table 11 Frequency of occurrence of ignition sources (reproduced from [47])

Ignition source	Hydrogen incidents		Non-hydrogen incidents	
	Number	%	Number	%
Arson	0	0	37	2.6
Collison	2	2.5	121	8.4
Flame	3	3.7	113	7.9
Hot surface	2	2.5	56	3.9
Electric	2	2.5	114	7.9
Friction spark	2	2.5	33	2.3
Not identified	70	86.3	942	64.5
Non-ignition	0	0	21	1.5
Total	81	100	1437	100

## Hydrogen releases

81 incidents all led to ignition

4 cases with delay between release and ignition

11 cases where the sources of ignition

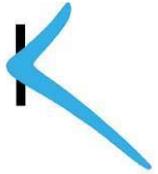
86.3% of incidents source not identified

## No-hydrogen releases

1.5% did not ignite

64.5% of ignition sources were not identified.

# Several mechanisms which could ignite accidentally released hydrogen



- ~~Reverse Joule-Thomson effect~~
- **Electrostatic ignition**
- **Spark discharges from isolated conductors**
- **Brush discharges**
- **Corona discharges**
- ~~Diffusion ignition~~
- ~~Sudden adiabatic compression~~
- ~~Hot surface ignition~~

Some form of electrostatic charging is a possible mechanism where spontaneous ignition of hydrogen leaks from high pressure occurred at ambient temperature

# Several mechanisms which could ignite accidentally released hydrogen

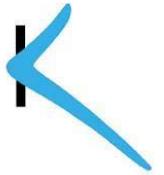
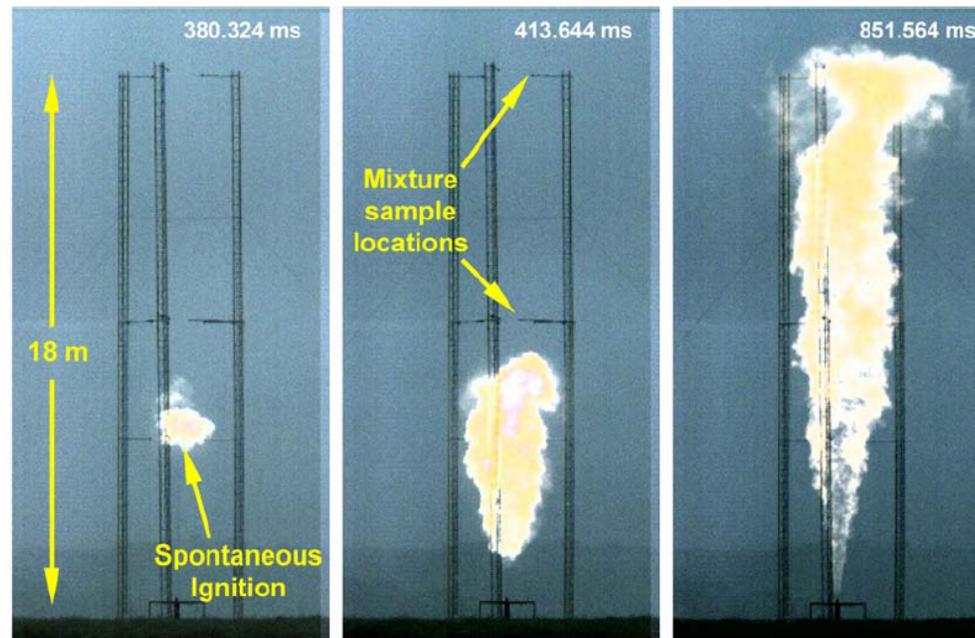
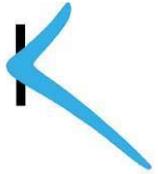


Fig. 32 High speed video frames from large-scale release experiments (reproduced from [46])

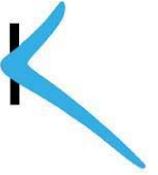


# Summary



- **Hydrogen fire burns quickly**
- **Hydrogen flames have low radiant heat**
- **Hydrogen fires are almost invisible**
- **Vertically oriented hydrogen jet fires**
  - Flame length: Delichatsios's correlation is valid for vertical turbulent hydrogen jets, both subsonic and choked (validation up to 172 bar)
  - Lift-off height: Kalghatgi's correlation can cover flames from subsonic jets to those from under-expanded sonic jets.
  - Radiative properties of hydrogen jet fires
    - the functional dependence for radiative heat flux established previously for a range of hydrocarbon flames is also applicable to hydrogen-jet flames.
    - the radiative fraction of large-scale jet flames is smaller than that predicted by the correlation obtained for laboratory-scale flames.
    - the optical thickness increases with the flame size
    - the radiative fraction for H<sub>2</sub> flames is nearly a factor of two lower than non-sooting hydrocarbon flames

# Summary



- **Horizontally oriented high pressure hydrogen jet flames**
  - Some tests have been carried out and some are still ongoing.
  - Little quantified information can be found in the public domain.
  - It needs to be verified whether the aforementioned findings on vertically oriented jet flames are applicable to horizontal flames.
- **Ignition of hydrogen leaks**
  - Some form of electrostatic charging is a possible mechanism where spontaneous ignition of hydrogen leaks from high pressure occurred at ambient temperature.